



Computational Modeling of Hypersonic Nonequilibrium Gas and Surface Interactions

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Grant FA9550-11-1-0309

Gas Phase

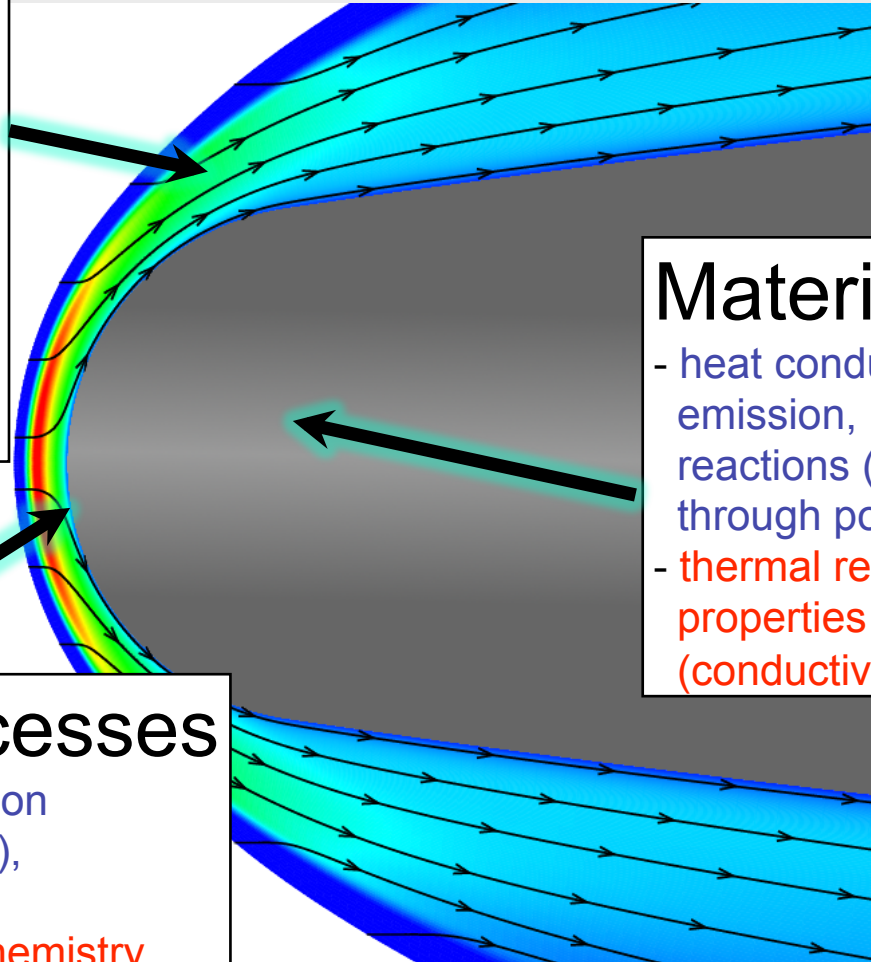
- strong shocks, thermochemical nonequilibrium, boundary layer, etc.
- CFD, relaxation times, Arrhenius rate coefficients with two-temperature model

Surface Processes

- accommodation, ablation (oxidation, sublimation), catalysis, melting, etc.
- coefficients, surface chemistry mechanism and rates

Material Response

- heat conduction, radiative emission, internal chemical reactions (pyrolysis), gas flow through porous media, etc.
- thermal response model, physical properties of complex materials (conductivity, emissivity...)



Project Goals



- Nonequilibrium gas-phase processes:
 - use computational chemistry and Master Equation analysis to perform detailed studies of:
 - thermal relaxation processes (T-R-V)
 - chemical processes (dissociation, exchange)
 - develop reduced order models for use in CFD
- Nonequilibrium gas-surface processes:
 - use coupled CFD-surface chemistry-material response tools to study gas-surface interactions (e.g., catalysis, ablation)
 - assess models using experimental data (flow and surface) generated in high-enthalpy facility (Fletcher, Univ. Vermont)



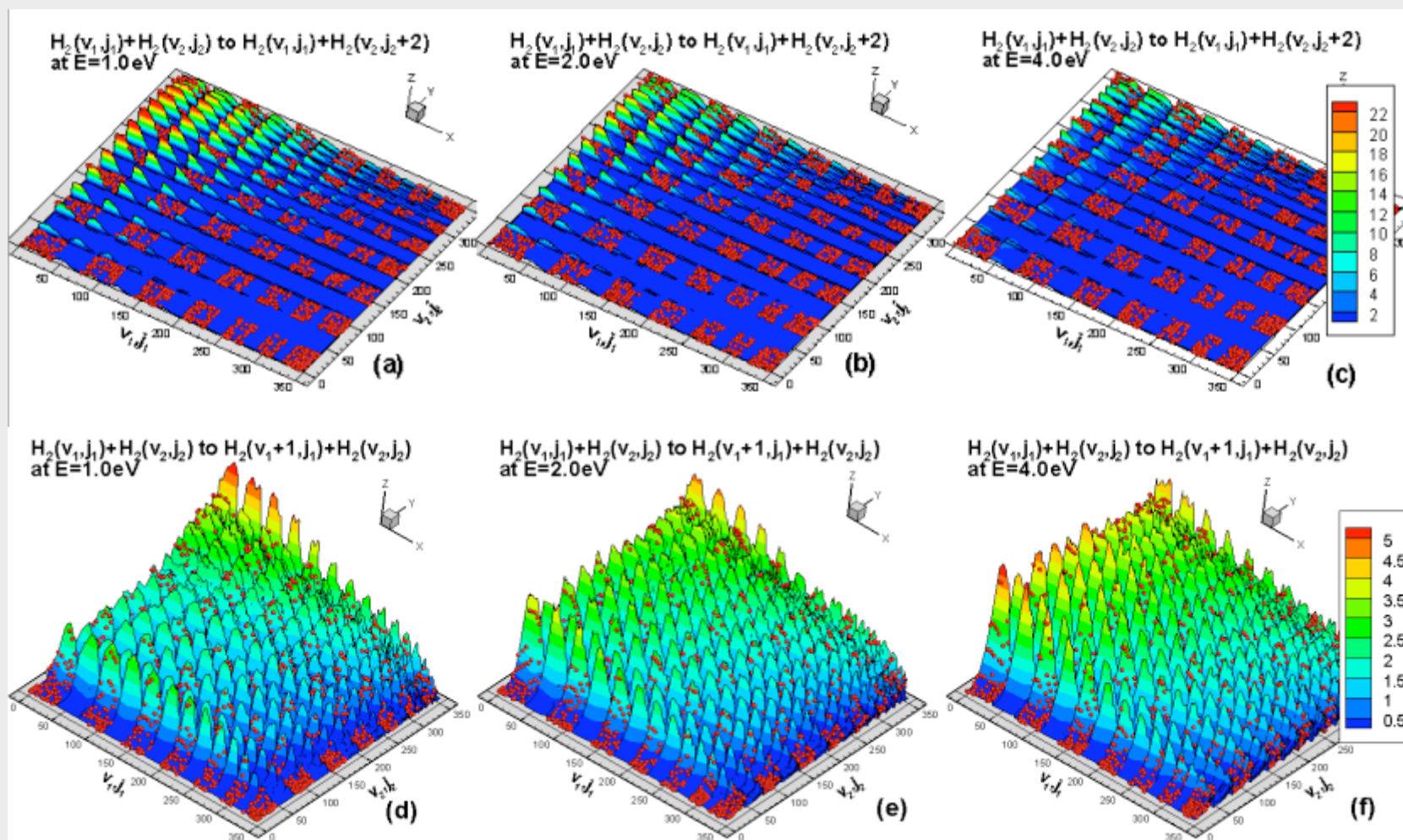
Gas Phase Studies: Technical Approach



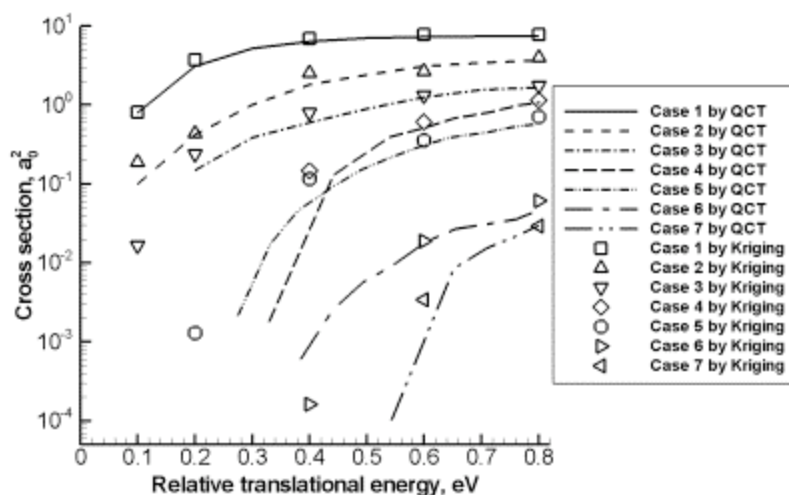
- State-to-state transition cross sections and rate coefficients:
 - compute data for all ro–vibrational states, e.g. using QCT
 - reduce the number of state-to-state transition rates evaluated using a response surface design technique (Kriging)
- Master Equation (ME) analysis of thermochemical relaxation:
 - constructed using complete sets of state-resolved transition rates for bound-bound and bound-free processes
 - compare results with existing measurements
 - use results to develop reduced-order thermochemistry models that can be implemented in CFD

Results: Bound-Bound H_2 Transitions

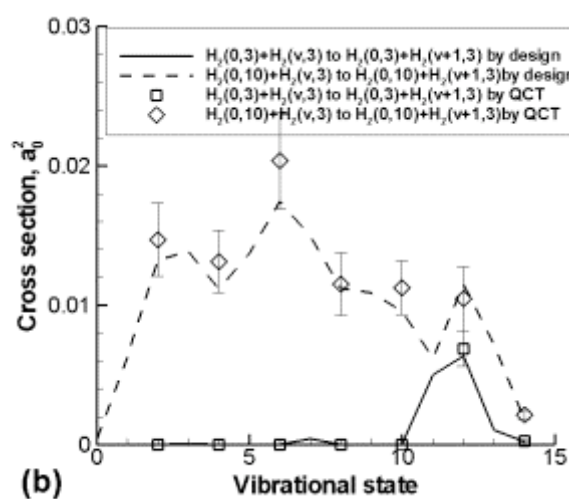
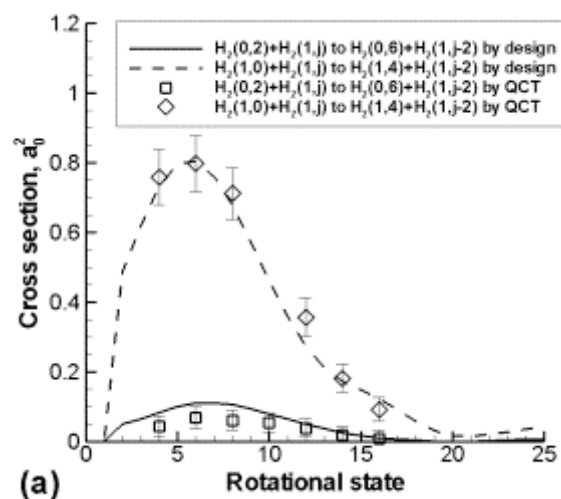
State-to-state cross sections obtained using response surface design method



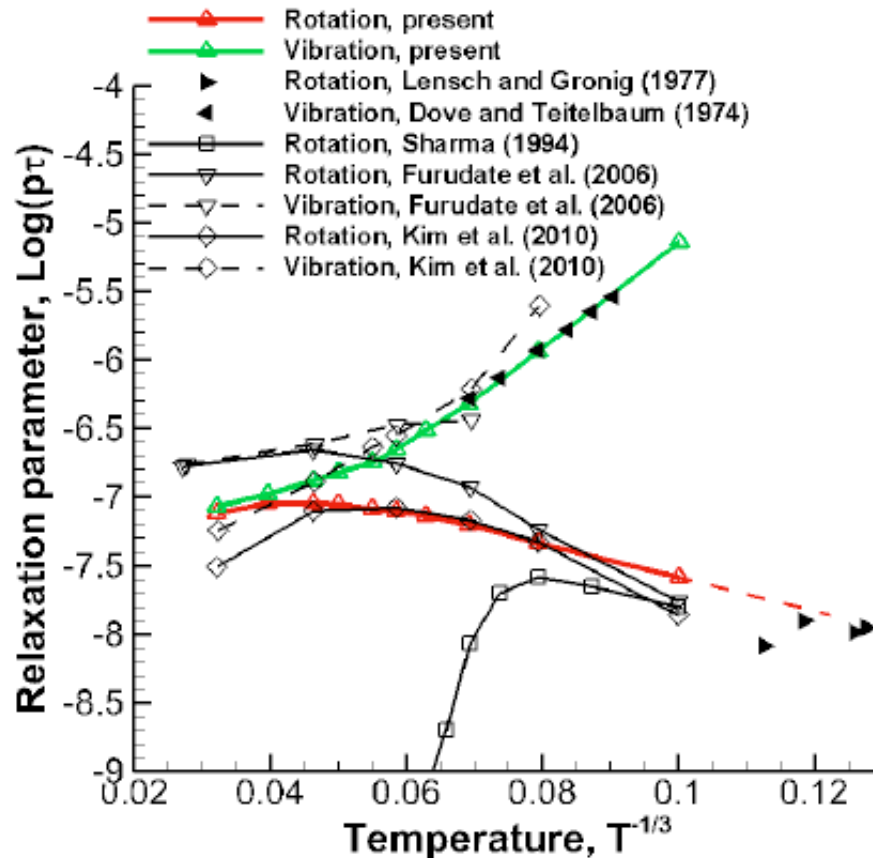
Results: Bound-Bound H_2 Transitions



State-to-state cross sections obtained using response surface design method calibrated using a small number of QCT evaluations (1,800 instead of 60,000!)



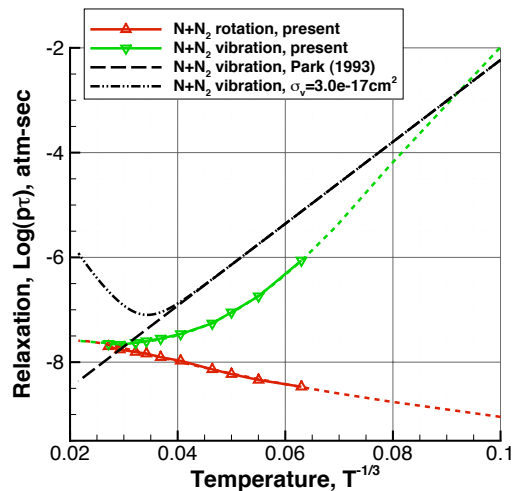
Results: H₂ Thermal Relaxation



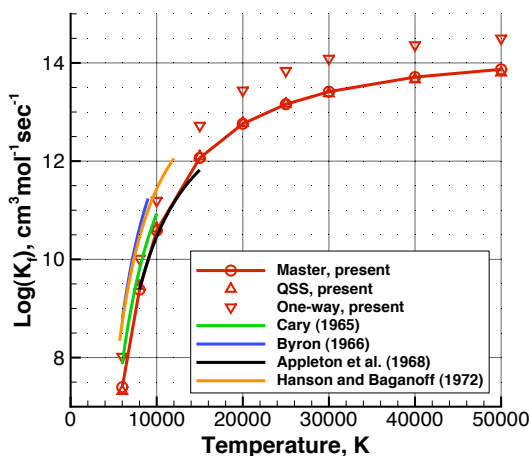
- Global relaxation parameters of the rotational and vibrational modes for H₂+H₂
- Rotational and vibrational relaxation times become similar at high temperature
- Technical details: Kim & Boyd, AIAA-2012-0362, Jan. 2012.

Analysis of N₂-N: Heat Bath Studies

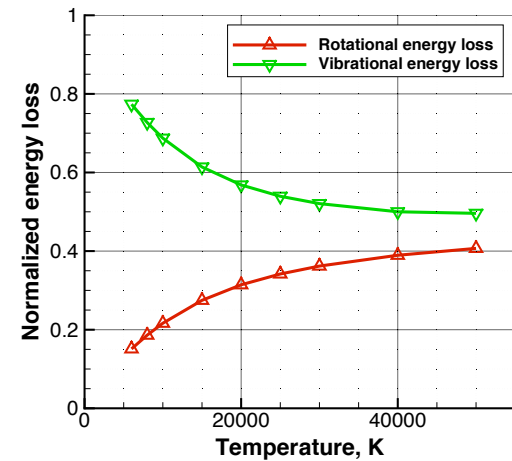
- State-to-state transition cross sections and rate coefficients:
 - use database of cross sections computed by Jaffe et al, NASA ARC
 - ME analysis involves solution of 9,390 equations
 - technical details: Kim & Boyd, AIAA-2012-2991, June 2012



Thermal Relaxation
Parameters



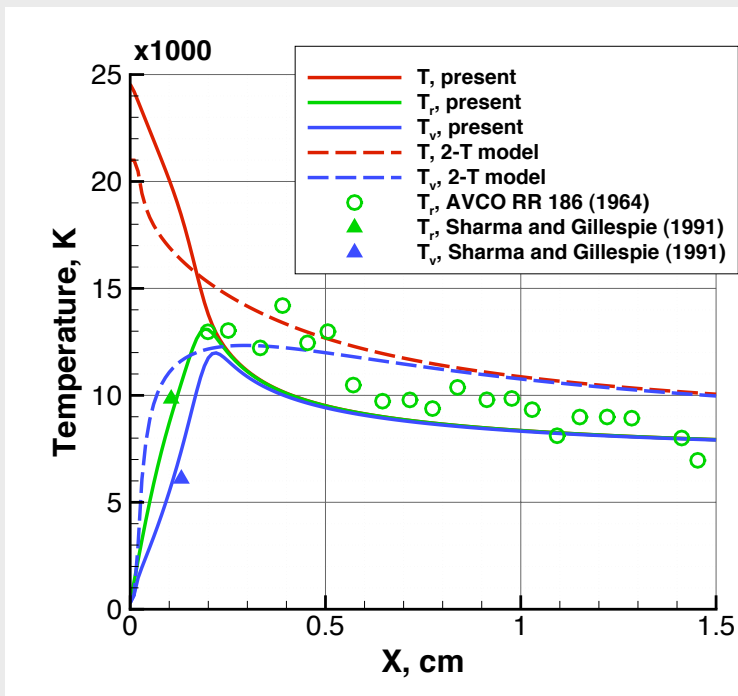
Chemical Reaction
Rates



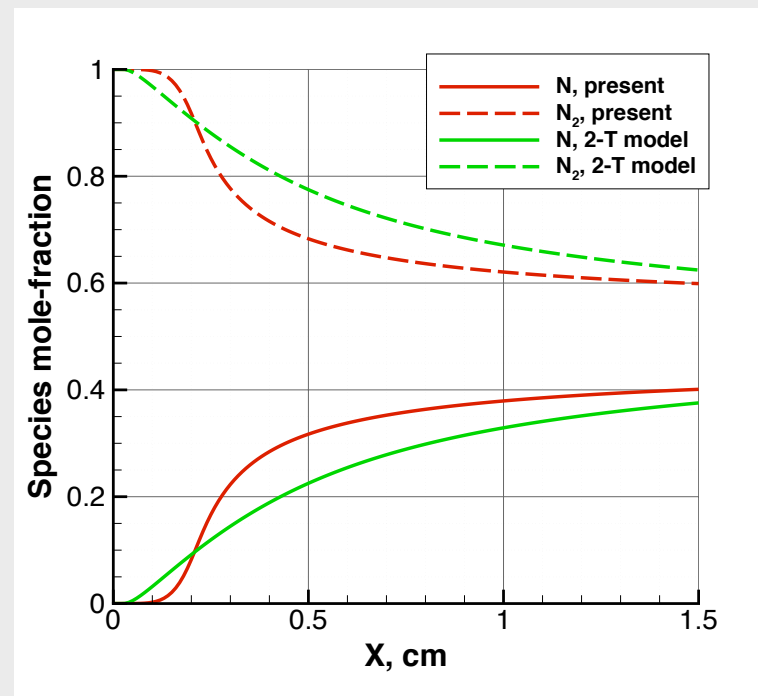
Energy Removal Due to
Chemistry

Analysis of N₂-N: Shock Tube Studies

- One-dimensional flow equations combined with Master Equation:
 - N₂-N₂ included macroscopically using standard models
 - applied to experiment of AVCO / Sharma
 - technical details: Kim & Boyd, AIAA-2012-2991, June 2012



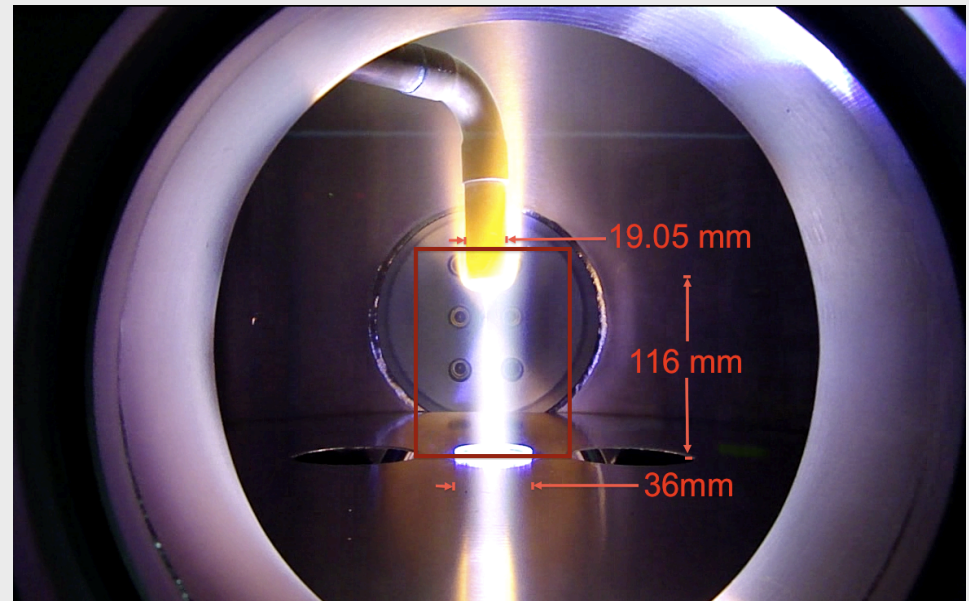
Temperature Profiles



Species Profiles

Gas-Surface Interactions: Assessment of Computations

- Collaboration with Prof. Doug Fletcher (UVM):
 - 30 kW Inductively Coupled Plasma (ICP) Torch Facility
- Samples exposed to high enthalpy gas flows
- Flow quantities measured using two-photon LIF:
 - N-atom number density
 - translational temperature
- Surface temperature and sample ablation also quantified



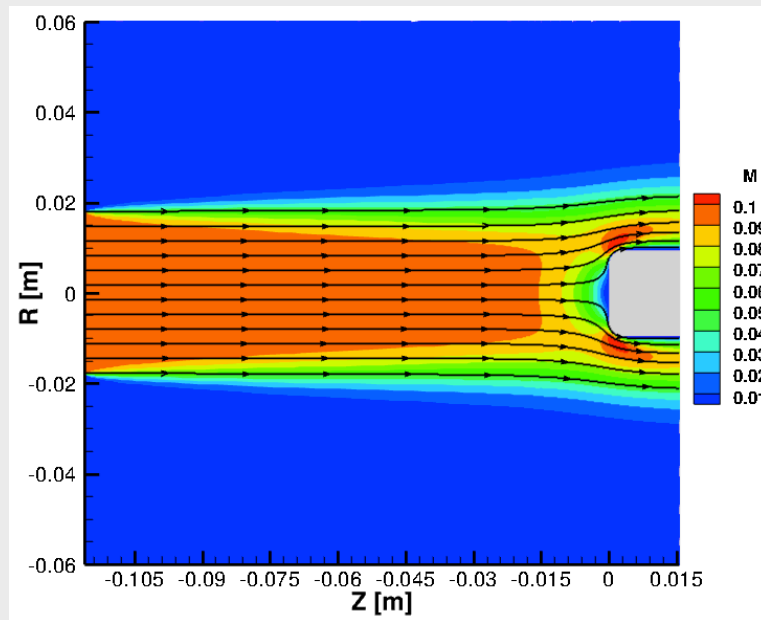
Graphite sample in nitrogen flow
(section in box is the portion simulated)
Source: Prof. D.G. Fletcher

Gas-Surface Interactions: Conditions Investigated

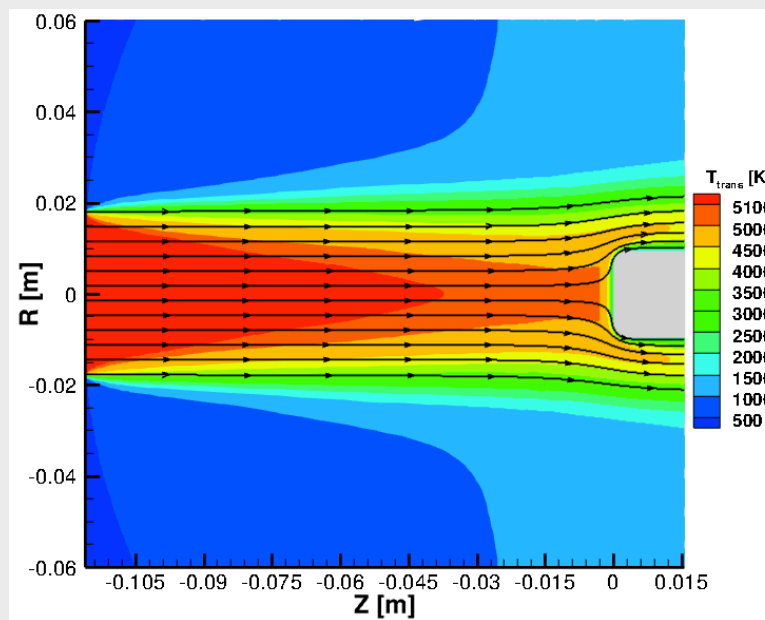
Free Stream:

Mass flow rate [kg/s]	Temperature T_∞ [K]	Pressure [kPa]	Wall temperature T_w [K]
0.001	5133	12	1590

Progress: subsonic inlet/outlet BCs added to LeMANS
 sensitivity to various thermochemistry models



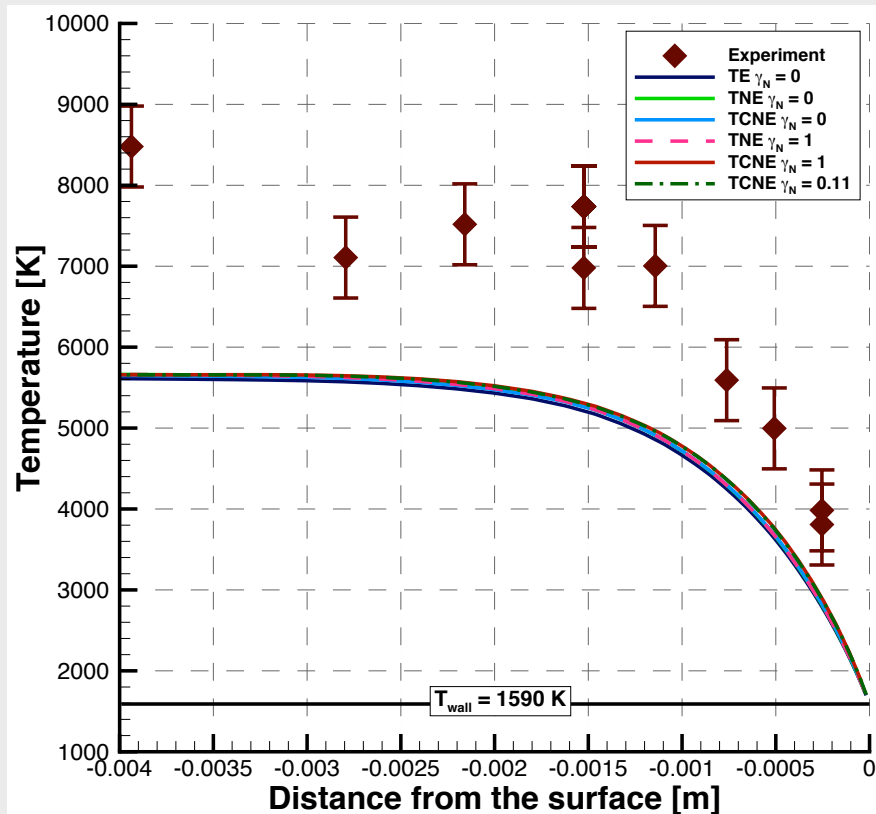
Mach number



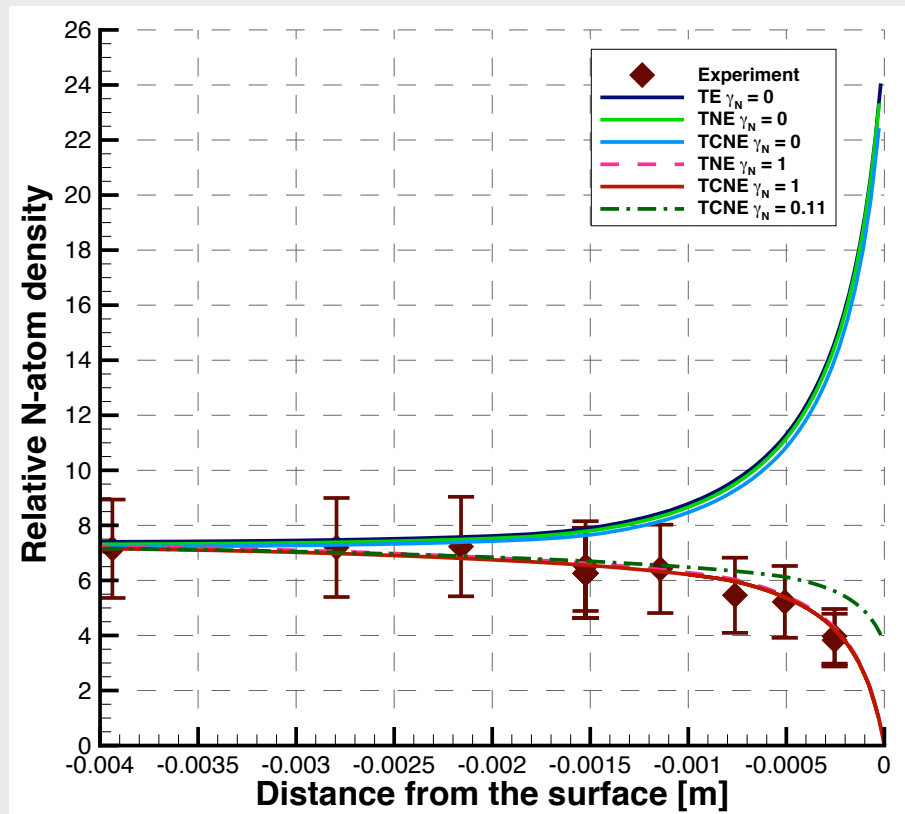
Translational temperature

Gas-Surface Interactions: Comparisons with Experiment

Comparisons along the stagnation streamline



Translational temperature



Relative N-atom density

Future Plans

- Nonequilibrium gas-phase processes:
 - develop T-R-V relaxation models for CFD from ME results
 - high fidelity CFD chemical reaction models including rotational mode will also be developed from ME analysis
 - continue analysis for other important air interactions
 - evaluation using existing experimental data sets
- Nonequilibrium gas-surface processes:
 - compare surface chemistry models (catalytic recombination, finite rate chemistry module of MacLean & Marschall)
 - model surface recession (material response code: MOPAR)
 - study sensitivity to gas-thermochemistry rates and models
 - assess modeling using Univ. Vermont experimental measurements of flow field properties and sample mass loss

Technical Challenges

- Nonequilibrium gas-phase processes:
 - large number of different air species interactions ($\text{N}_2\text{-M}$, $\text{O}_2\text{-M}$, NO-M , etc.)
 - fidelity required from computational chemistry?
 - Master Equation analysis becoming expensive
 - lack of modern, validation quality, experimental data
- Nonequilibrium gas-surface processes:
 - isolating contributions of competing mechanisms to effects observed (e.g. flow processes, catalysis, ablation)
 - uncertainties in facility operation (e.g. ICP exit conditions)

Technical Approach: Computational Tools



- **LeMANS**

(Scalabrin and Boyd: AIAA-2006-3773)

- Navier-Stokes CFD code
- finite volume FVS
- implicit time integration (point/line)
- 2D/3D unstructured mesh
- parallel, domain decomposition
- finite rate thermo-chemical nonequilibrium effects
- validated for hypersonic flow using experiments, codes

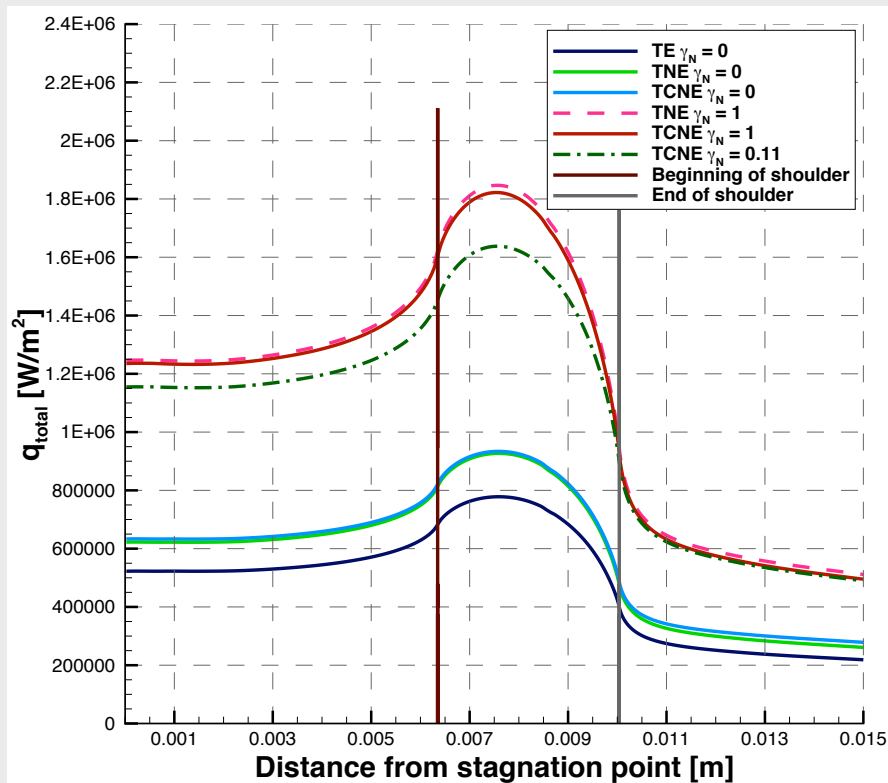
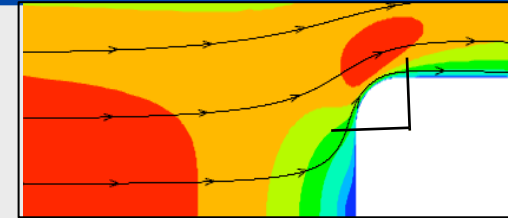
- **MOPAR**

(Martin and Boyd: AIAA-2009-3597)

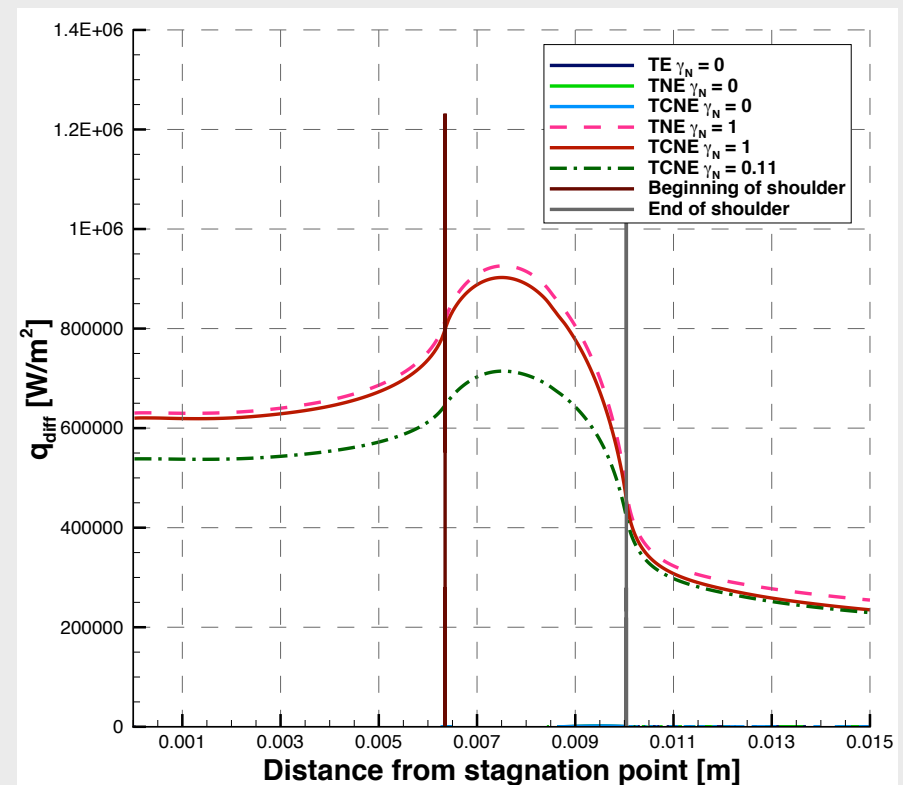
- material response code
- control volume finite element (CVFEM)
- quasi-1D
- pyrolyzing/non-pyrolyzing ablators
- momentum conservation through Darcy's Law (or Forcheimer's Law)
- moving boundaries
- has been coupled to LeMANS

Results: Surface Properties

Total Heat flux = (Translational + Vibrational)
convective heat flux +
Diffusive heat flux



Total heat flux



Diffusive heat flux